



STUDIES ON EXTRACTION, PHYTOCHEMICALS AND THIN LAYER CHROMATOGRAPHIC (TLC) ANALYSIS OF STEM BARK AND ROOT EXTRACTS OF *MORINDA LUCIDA* (NVUVU)

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Abstract: The present study is aimed at qualitatively screening for phytochemicals in *M. lucida* stem bark and root to determine the major classes of compound present in them. This was done using three different solvents successively to extract. The effects of solvent-to-solid ratio on the percent yield of extracts were assessed. Phytochemical analysis was performed by simple chemical analysis including thin layer chromatographic (TLC) analysis which was carried out using *n*-hexane, ethyl acetate (3:1), *n*-hexane, ethyl acetate (3:2) and *n*-hexane, ethyl acetate, and methanol (2:1:1) solvent systems. The optimal percent yield as a factor of solvent-to-solid ratio of the extracts obtained is 10:1. The phytochemical analysis revealed the presence of saponins, tannins, anthracenes, flavonoids, alkaloids, carbohydrate, gum and Mucilages and protein. The TLC analysis of *M. lucida* stem bark revealed the presence of 8 spots using *n*-hexane, ethyl acetate (3:1) solvent system and 7 spots using *n*-hexane, ethyl acetate (3:2) solvent system. The TLC results of the root extracts shows 10 spots using *n*-hexane, ethyl acetate (3:1) solvent system, 8 spots using *n*-hexane, ethyl acetate (3:2) solvent system and 2 spots using *n*-hexane, ethyl acetate, and methanol (2:1:1) solvent system. The present study provides evidence that solvent extract of *M. lucida* contains some possible medicinally important bioactive compounds and this justifies the use of the plant species as traditional medicine for treatment of various diseases.

Keywords: Phytochemicals, solvent-to-solid ratio, thin-layer chromatography, *Morinda lucida*, Successive extraction.

Introduction

Phytochemicals are naturally occurring non-nutritive chemicals produced by plants for their protection, recent research demonstrate that they can also protect humans

against diseases. That they have protective or disease preventive properties such as antioxidant, enzymes stimulants; have hormonal action, anti-bacterial activity and DNA replication action [Sowemimo et al., 2007] that

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acts as natural defense system for the host plants against bacteria, fungi and viruses [Njerua et al., 2013], [Abhijeet et al., 2018] and in addition provide colour, aroma and flavor [Sangeetha et al 2014], [Mace, 1963]. These chemicals called phytochemicals give the plant color (plant pigments), aroma and flavor. Some of the most important phytochemicals are carotenes, chlorophyll, and flavonoids and are naturally present in many foods [Olaniyan, 2016]. Phytochemicals help to protect our cells against oxidative damage and reduce the risk of developing certain types of cancer, reduce the risk of high blood pressure responsible for heart failure, and stimulate enzymes that make the estrogen less effective and could reduce the risk for breast cancer, prevent the multiplication of cancer cells by protecting DNA from carcinogens. They also have antibacterial properties, reduce the risk of urinary tract infections and will improve dental health, prevention and treatment of colorectal cancer, diabetes, heart disease, high cholesterol, gastrointestinal disorders and obesity. Some help to alleviate cholesterol levels in patients with cardiovascular complications, and may also improve dental health, help improve your overall health and boost immune system [Olaniyan, 2016].

Materials and methods

Sample Collection and Preparation

The sample *Morinda lucida* (Nvuvu) stem back and roots were collected from Oyofe Oghe in Ezeagu Local Government Area on 29th June, 2019. Oyofe Oghe lies Northwest of Enugu Metropolis. This plant parts were collected by following leads supplied by local healers in Oghe communities where the plants are used in traditional medicine. The sample *M. lucida* stem back and roots were air dried in-doors to a constant weight at room temperature [Dilika et al., 1996], [Baris et al 2006] as differences in water content may affect solubility on subsequent separation by liquid-liquid extraction. The samples were

ground into fine powder as this will increase the surface area for extraction thereby increasing the rate of extraction.

Successive extraction

Soxhlet extraction method was used in the extraction process. The solvent-to-sample ratio affects the quantity and quality of constituents obtained. In this studies solvent to sample ratios of 5:1, 10: 1, 15:1, 20:1 and 25:1 (v/w) solvent to dry weight ratio were used. Thirty grams of dried ground plant material (stem back and roots) were weighed and transferred into a 25 x 100 mm cotton extraction thimble and exhaustively extracted with 150, 300, 450, 600 mL analytical grade solvents of increasing polarity to ensure extraction of a wide polarity range of compounds. The extracts were concentrated to a constant weight and labeled n-hexane, ethyl acetate and methanol extracts of *M. lucida* stem back and roots respectively [Green, 2004], [Das et al ., 2010], [Odebiyi and Sofowara, 1978] - [Ogbuanu et al., 2014]

Screening for phytochemicals in the plant extracts

The extracts were subjected to a phytochemical screening using the following tests: the Wagner [35], Dragendorff [Sofowara, 1993], - [Trease and Evans, 1989] Mayer [Trease and Evans, 1989] and Krait tests for alkaloids; the Ninhydrin [Yasuma and Ichikawa, 1953], [Devi et al., 2020], [Hemantkumar, 2017] Biuret [Sofowara, 1993],[Priyanga et al., 2014] and Millon [25] tests for amino acids and proteins; Fehling and Benedict tests for carbohydrates [Priyanga et al., 2014] and Molish [Sofowara, 1993] spot and saponification tests for Fixed oils and Fats [Kokate, 1999], Borntrager [Evans, 1997], Legal, Fehling and Ferric chloride [Sofowara, 1993], tests for glycosides; Born Tragers test for anthracenes [Sofowara, 1993],_the ferric chloride [Mace, 1963] and gelatin [Evans, 1997] tests for tannins; the Shinoda, alkaline and lead acetate tests [20] tests for flavonoids; the Libermann-Burchard test for phytosterols [Olaniyan, 2016] the Rosenthaler test for saponins [Sofowara, 1993], [Kokate, 1999], the absolute alcohol test for gum and



Mucilages [Sai et al., 2011], [Whistler and BeMiller, 1993].

Thin layer chromatographic analysis (TLC)

Thin layer chromatographic analysis is often used in evaluating medicinal plants material [WHO, 1998.]. The ascending technique was employed in the TLC analysis. A clean dry chromatographic tank with 50ml of the running solvent (mobile phase) made-up of n-hexane and ethyl acetate (3:1), (3:2) and n-hexane, ethyl acetate and methanol (2.1.1) respectively were used to develop the chromatograms. The spots on the TLC plate after development were visualized in iodine tank and the position of the spots marked with pencil. The retention factor (R_F values) for each spot was calculated for each extract.

Results and discussion

If the extraction is for general phytochemical analysis or screening then the larger the variety of compounds the

solvent will extract the better, because there is a better chance that biologically active compounds will be present [Eloff, 1998]. The results showed that n-hexane is a better extracting solvent followed by ethyl acetate as they extract more variety of compounds respectively than methanol. Methanol extracted fewer secondary metabolites and probably will contain traces of residual solvent that is toxic and likely will interfere with the bioassay. Due to low toxicity, ease of evaporation at low heat and ease of subsequent handling of the extracts, n-hexane followed by ethyl acetate are recommended for extraction of secondary metabolites in *M. lucida* stem bark and root extracts.

Fig. 1 shows plots for the percent maximum yield as a function of solvent-to-solid ratios ranging from 5:1 to 25:1 for both stem bark and root extractions.

Table 1: Effect of solvent to sample ratios on percent yield of *Morinda lucida* (Nvuvu) stem bark and root extracts.

Extract	Stem bark					Root				
	% Yield solvent to sample ratios					% Yield solvent to sample ratios				
	5:1	10:1	15:1	20:1	25:1	5:1	10:1	15:1	20:1	25:1
n-Hexane	0.248	0.484	0.484	0.484	0.484	0.258	0.494	0.494	0.494	0.494
Ethyl acetate	0.239	0.468	0.468	0.468	0.468	0.301	0.478	0.478	0.478	0.478
Methanol	0.201	0.233	0.233	0.232	0.233	0.213	0.243	0.243	0.242	0.243

The data presented correspond to the maximum values extracted by each ratio. As expected the solvent-to- solid ratio was optimal at 10:1 ratio. In this studies (Table 1) solvent-to-dry sample ratios of 10: 1 (v/w) used yielded more and this agreed as reported as ideal by [Ncube et al 2008].

Out of 30 g of plant material, optimal yield obtained for stem bark and root was at a ratio of 10:1 solvent–to-solid for all the extracting solvent (Table 1). In this successive

exhaustive extraction, methanol gave the least percent yield of extract. This probably is due to the fact that most compounds present in the plant samples are soluble in n-hexane and ethyl acetate.

The results of TLC analysis of *M. lucida* stem bark (Table 3) shows a total of 8 spots using n-hexane, ethyl acetate (3:1) solvent system (3 spots for n-hexane extract, 5 spots for ethyl acetate extract and non for methanol extract respectively). For n-hexane and ethyl acetate (3:2) solvent



system revealed a total of 7 spots (n-hexane and ethyl acetate extracts gave 4 spots and 3 each and none for methanol). Solvent system n-hexane, ethyl acetate and methanol (2:1:1) gave no spot for n-hexane, ethyl acetate and methanol extracts.

The presence of anthracene only in the root extract of *M. lucida* plant may be responsible for the use of the root in Oghese traditional medicine. This may probably be working in synergy with other secondary metabolites and probably justifying the use of root in traditional medicine.

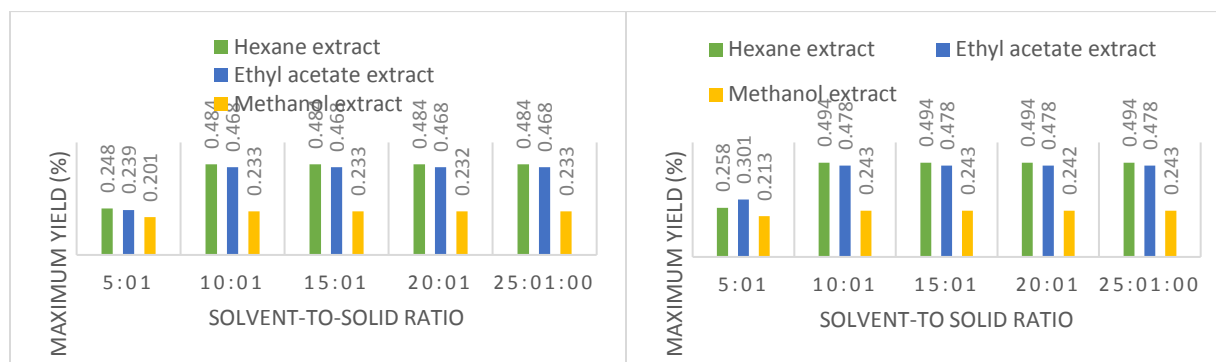


Fig. 1: Maximum percent extraction yield values of stem bark and root estimated with different solvent-to-solid ratio

The results of phytochemical analysis of the n-Hexane, ethyl acetate and methanol extracts of the stem bark and root of *M. lucida* plant were summarized in Table 2. The result showed that the plant is rich in saponins, tannins, anthracenes, flavonoids, alkaloids, carbohydrate and protein. The presence of these constituents in the investigated plant leaves account for their usefulness in traditional medicine. The degree of precipitation of secondary metabolites varies in solvents. This probably

may be due to varying degree of solubility of different phytoconstituents in different solvents [Sangha]. n-Hexane (a non-polar solvent with dielectric constant of 1.9) stem bark extract shows the presence of saponins, hydrolysable tannins alkaloids, flavonoids, carbohydrate and proteins while the root contain saponins, tannins, anthracenes, alkaloids, flavonoids, carbohydrate and proteins.

Table 2: Results of phytochemical screening of *Morinda lucida* (Nvuvu) plant stem back and roots extracts

Parameters	Stem back			Root		
	A	B	C	A	B	C
Saponin	+	+	+	+	+	+
Saponin Glycoside	+	+	+	-	-	-
Tannins	-	-	-	-	-	-
Hydrolysable Tannins	+	-	-	+	+	-
Pseudo Tannin	-	-	-	-	-	-



Test of Tannin (Using Gelatin)	-	-	-	+	-	-
Digital Glycoside	-	-	+	-	-	-
Glycoside General	-	-	-	-	-	-
Anthracene (Born Tragger Test)	-	-	-	+	-	-
Resins	-	-	-	-	-	-
Steroid and Triterpenoids (Libemann Buchard) Test	-	-	+	-	-	-
Volatile Oil Test	-	-	-	-	-	-
Alkaloid: (a). Dragendoff Test	-	-	-	-	-	-
(b). Wagner's Test	-	-	-	-	-	-
(c). Mayer's Test	-	-	-	-	-	-
(d). Krait's Test	+	+	+	+	+	+
Flavonoid Test (a) Magnesium ribbon test				-	-	+
(b). Alkaline test	+	-	-	-	-	-
(c). Lead Acetate test	-	+	-	-	+	+
Carbohydrate Test	+	-	+	+	-	+
Protein	+	-	+	+	+	+
Gum and mucilages	-	-	-	-	+	-

+ = positive; - = negative; A = n-Hexane; B = Ethyl acetate; C = Methanol

Ethyl acetate (slightly polar solvent with dielectric constant of 6.4) stem bark extract gave saponins, hydrolysable tannins, alkaloids and flavonoids while the root extract gave saponins, hydrolysable tannins, alkaloids, flavonoids and protein. Methanol (polar solvent with dielectric constant of 31.0) stem bark extract gave saponin, saponin glycoside, alkaloids, carbohydrate and protein while the root extract contain saponin, alkaloids, flavonoids, carbohydrate and protein.

The TLC results of the root extracts shows a total of 10 spots (6 spots for n-hexane and 4 spots for ethyl acetate

extracts while non for methanol extract) using n-hexane, ethyl acetate (3:1) solvent system. The results also using n-hexane, ethyl acetate (3:2) solvent system revealed a total of 8 spots (4 spots each for n-hexane and ethyl acetate extracts) while non for methanol extract. Hexane, ethyl acetate, and methanol (2:1:1) solvent system gave One spot each for n-hexane and ethyl acetate extracts respectively and non for methanol extract.

TABLE 3: The number of spots for TLC analysis of *Morinda lucida* (Nvuvu) stem back and root extracts.

Solvent system	Stem back			Root		
	A	B	C	A	B	C
n-hexane, ethyl acetate (3:1)	3	5	-	6	4	-



n-hexane, ethyl acetate (3:2)	4	3	-	4	4	-
n-hexane, ethyl acetate, methanol (2:1:1)	-	-	-	1	1	-

A = n-Hexane; B = Ethyl acetate; C = Methanol

Conclusion

The solvent-to-solid ratio was optimal at 10:1 ratio, meaning that more extract will be obtained at this ratio. The results obtained in the present study indicated *M. lucida* stem bark and root as a rich source of bioactive components. The results of TLC analysis of *M. lucida* stem bark shows a total of 8 spots using n-hexane, ethyl acetate (3:1) solvent system and 7 spots using n-hexane, ethyl acetate (3:2) solvent system. Solvent system n-hexane, ethyl acetate and methanol (2:1:1) gave no spot for n-hexane, ethyl acetate and methanol extracts. The n-hexane, ethyl acetate (3:1) is a better solvent system followed by n-hexane, ethyl acetate (3:2). While n-hexane, ethyl acetate and methanol (2:1:1) solvent system could not partition the phytochemical and as such not a good solvent system.

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